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Key indicators

Single-crystal X-ray study T = 173 K Mean σ (C–C) = 0.002 Å R factor = 0.037 wR factor = 0.114 Data-to-parameter ratio = 17.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. The carboxyl and carboxymethyl substituents in the title compound, $C_{12}H_{12}O_4$, lie on opposite sides of the cyclopropyl ring and each carboxyl group participates in a centrosymmetric hydrogen-bonding scheme, typical for carboxylic acids, leading to the formation of zigzag chains in the crystal.

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organic papers

Comment

The title compound, (I), was prepared by the controlled hydrolysis of the parent methyl diester (Avery et al., 2001). The structure determination confirms the overall stereochemistry, notably the positioning of the substituents around the cyclopropyl ring as shown in Fig. 1. The plane of the C11/O11/O11A carboxylic acid group is almost orthogonal to the cyclopropane plane, as seen in the value of the X-C1-C11-O11torsion angle of 4.1 (2)°, where X is the midpoint of the C2– C3 bond. By contrast, the dihedral angle formed by the planes of the phenyl and cyclopropyl rings is 68.70 (11)°. The conformation is thus described as cis-bisected in accord with Allen (1980). The distal C2-C3 bond distance of 1.5027 (19) Å is shorter, as expected (Allen, 1980), than the respective vicinal C1-C2 and C1-C3 bond distances of 1.5196 (19) and 1.5195 (17) Å, reflecting the influence of the π -acceptor carboxylic acid group upon the C–C bonds within the cyclopropyl ring.



Molecules associate in the lattice *via* hydrogen-bonding interactions involving both carboxylic acid residues so that each carboxylic acid group associates with its centrosymmetrically related mate. Thus, O-H11A = 0.92 Å, $H11A \cdots O11^{i} = 1.74$ Å and $O11A \cdots O11^{i} = 2.653$ (2) Å, and the angle at $H11A = 174^{\circ}$ [symmetry code: (i) 1 - x, -1 - y, -z]. The comparable parameters for H22A and O22ⁱⁱ are 0.87, 1.78 and 2.646 (2) Å, and 175° [symmetry code: (ii) 2 - x, -y, -z]. Such association leads to the formation of zigzag chains extending in the *ab* plane.

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Figure 1

The molecular structure and crystallographic numbering scheme for (I). Displacement ellipsoids are shown at the 50% probability level (Johnson, 1976).

Experimental

The title compound was prepared according to the literature procedure of Avery et al. (2001). Colourless crystals were obtained by slow evaporation of a dichloromethane/heptane solution of the compound (m.p. 458-460 K).

Crystal data

 $C_{12}H_{12}O_4$ $M_r = 220.22$ Monoclinic, $P2_1/n$ a = 8.555 (4) Å b = 8.814 (4) Å c = 14.849(2) Å $\beta = 101.70 \ (2)^{\circ}$ V = 1096.5 (6) Å³ Z = 4 $D_x = 1.334 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation Cell parameters from 24 reflections $\theta = 7.3 - 10.7$ $\mu = 0.10 \text{ mm}^{-1}$ T = 173 KBlock, colourless $0.45 \times 0.29 \times 0.16 \ \mathrm{mm}$

Data collection

Rigaku AFC-7 <i>R</i> diffractometer ω -2 θ scans 4704 measured reflections 2510 independent reflections	$h = -11 \rightarrow 6$ $k = 0 \rightarrow 11$ $l = -19 \rightarrow 19$ 3 standard reflections
1829 reflections with $I > 2\sigma(I)$ $R_{int} = 0.020$ $\theta_{max} = 27.5^{\circ}$	every 400 reflections intensity decay: 0.3%
Refinement	
Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.037$	$w = 1/[\sigma^2(F_o^2) + (0.0728P)^2 + 0.3498P]$
$wR(F^2) = 0.114$	where $P = (F_o^2 + 2F_c^2)/3$
S = 0.90	$(\Delta/\sigma)_{\rm max} < 0.001$
2510 reflections	$\Delta \rho_{\rm max} = 0.21 \ {\rm e} \ {\rm A}^{-3}$
146 parameters	$\Delta \rho_{\rm min} = -0.23 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

The C-bound H atoms were placed in geometrically calculated positions and included in the final refinement in the riding-model approximation with an overall isotropic displacement parameter. The H atoms on O atoms were located from a difference map but were not refined.

Data collection: MSC/AFC Diffractometer Control Software (Molecular Structure Corporation, 1996); cell refinement: MSC/AFC Diffractometer Control Software; data reduction: TEXSAN (Molecular Structure Corporation, 1997); program(s) used to solve structure: SIR97 (Altomare et al., 1994); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); software used to prepare material for publication: SHELXL97 (Sheldrick, 1997).

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References

- Allen, F. H. (1980). Acta Cryst, B36, 81-96.
- Altomare, A., Cascarano, M., Giacovazzo, C., Guagliardi, A., Burla, M. C., Polidori, G. & Camalli, M. (1994). J. Appl. Cryst. 27, 435.
- Avery, T. D., Fallon, G., Greatrex, B. W., Pyke, S. M., Taylor, D. K. & Tiekink, E. R. T. (2001). J. Org. Chem. Submitted.
- Johnson, C. K. (1976). ORTEPII. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.

Molecular Structure Corporation (1996). MSC/AFC Diffractometer Control Software. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA.

Molecular Structure Corporation (1997). TEXSAN for Windows. Version 1.05. MSC, 3200 Research Forest Drive, The Woodlands, TX 77381, USA. Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.